# ARBORONE AND 7-OXO-DIHYDROGMELINOL: TWO NEW KETO-LIGNANS FROM GMELINA ARBOREA ${ }^{1}$ 

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ABSTRACT.—The structures of two new 2,3,4-trisubstituted tetrahydrofuran lignans 1a and 2a, isolated from the heartwood of Gmelina arborea, have been determined. In addition, two known furofuran lignans, paulownin acetate and epieudesmin, were isolated along with methyl trans-p-methoxycinnamate and trans-p-hydroxycinnamic acid. The conversion of 1a into arboreol is also reported.

Extraction of the heartwood of Gmelina arborea L. (1) has yielded a number of novel lignans, including arboreol, isoarboreol, methyl arboreol, gummadiol, and gmelanone $(2,3)$, the latter having a rearranged carbon skeleton derivable by acid-catalyzed rearrangement of arboreol (4). The present paper describes the isolation of six additional compounds of which two, arborone (1a) and 7-oxo-dihydrogmelinol (2a), are new. They are members of a relatively small group of lignans having aroyl-substituted tetrahydrofuran structures (5-7). The remaining four compounds comprise two known furofurans, paulownin acetate (8) and epieudesmin ( 9,10 ), in addition to methyl trans-$p$-methoxycinnamate and trans- $p$-hydroxycinnamic acid.

## RESULTS AND DISCUSSION

Arborone (1a), $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}_{8}$, mp 216-218,$[\alpha]^{28} \mathrm{D}+81.48\left(\mathrm{CHCl}_{3}\right)$, exhibited uv maxima at 206,234 , and $285 \mathrm{~nm}(\log \in 4.44,4.17$, and 4.01). It formed a 2,4 -dinitrophenylhydrazone, mp $268-270^{\circ}$, and on treatment with $\mathrm{Ac}_{2} \mathrm{O}$ and pyridine gave a diacetate (1b). Arborone gave a positive Labat test and in its ${ }^{1} \mathrm{H} \mathrm{nmr}$ and ms gave a clear indication of the presence of two 3,4-methylenedioxyphenyl groups. In its ir spectrum it gave a strong absorption at $1640 \mathrm{~cm}^{-1}$ and this, coupled with the appearance of two low field aromatic protons at $\delta 7.73 \mathrm{dd}(2,8)$ and $\delta 7.55 \mathrm{~d}(2)$ in the ${ }^{1} \mathrm{H}-\mathrm{nmr}$ spectrum, suggested the presence of an aroyl residue. This was supported by the ms, which gave as the base peak an ion having $m / z 149\left(\mathrm{ArCO}^{+}\right)$. The two-proton signal at $\delta 3.60$ in the ${ }^{1} \mathrm{H}-\mathrm{nmr}$ spectrum of arborone, which was shifted downfield to $\delta 4.03 \mathrm{dd}(6.8,11.7)$ and $4.09 \mathrm{dd}(7.5,11.7)$ in the spectrum of the acetate, was consistent with the presence

$\begin{array}{ll}\text { 1a } & \begin{array}{l}\mathrm{Ar}=\text { piperonyl } \\ \\ \mathrm{R}=\mathrm{H}\end{array} \\ \text { 1b } & \mathrm{Ar}=\text { piperonyl } \\ \mathrm{R}=\mathrm{Ac}\end{array}$


2a $\mathrm{Ar}^{1}=$ veratryl
$\mathrm{R}=\mathrm{H}$
2b $\quad \mathrm{Ar}^{1}=$ veratryl
$\mathrm{R}=\mathrm{Ac}$

$3 \mathrm{Ar}=$ piperonyl
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of a primary $\mathrm{CH}_{2} \mathrm{OH}$ group. Further analysis of the ${ }^{1} \mathrm{H}$-nmr spectra (Table 1) led to structure $\mathbf{1 a}$ for arborone, and this was supported by the ms, which contained a number of ions at $m / z 338,192,176,164$, and 161 that are characteristic of the arboreol series $(2,3)$. Indeed, treatment of arborone with HCl in ErOAc gave a product, $\mathrm{mp} 162^{\circ}$, which was identical in all respects with arboreol (3) (2), thus confirming the structure and stereochemistry of 1a. It seems likely that arborone arises in nature by hydrolytic cleavage of arboreol. The close relationship between the structures of the two compounds is confirmed by this experiment which thereby establishes the structure of arborone on a firm basis.

Table 1. ${ }^{1} \mathrm{H}-\mathrm{nm}$ Spectra ${ }^{\mathrm{a}}$

| Proton | Compounds |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $\begin{gathered} \text { Arborone (1a) } \\ \mathrm{CDCl}_{3}+\mathrm{DMSO}-d_{6} \\ (100 \mathrm{MHz}) \end{gathered}$ | $\begin{gathered} \text { Arborone } \\ \text { diacetate }(\mathbf{1 b}) \\ \mathrm{CDCl}_{3}(360 \mathrm{MHz}) \end{gathered}$ | $\begin{gathered} \text { 7-Oxo-dihydro } \\ \text { gmelinol (2a) } \\ \mathrm{CDCl}_{3}(100 \mathrm{MHz}) \end{gathered}$ | 7-Oxo-dihydrogmelinol acetate (2b) $\mathrm{CDCl}_{3}(360 \mathrm{MHz})$ |
| 2 | $4.67 \mathrm{~d}(5)^{\text {b }}$ | $4.65 \mathrm{~d}(5.0)$ | 5.09 s | 4.91 s |
| 3 | 2.7-2.9 m | $3.01 \mathrm{q}(6.8)$ | - | - |
| 4 | - | - | 4.29 t (7.1) | 4.31 t (7.3) |
| 5 | $3.96-4.6 \mathrm{~m}$ | $4.52 \mathrm{~d}(11.0)$ | $4.19 \mathrm{dd}(7.1,6.1)$ | $4.19 \mathrm{dd}(7.3,8.5)$ |
|  |  | 4.68 d(11.0) | 4.42 d (7.1) | $4.45 \mathrm{dd}(7.3,8.5)$ |
| $\mathrm{CH}_{2}$ | 3.60 m | $\begin{aligned} & 4.03 \mathrm{dd}(6.8,11.7) \\ & 4.09 \mathrm{dd}(7.5,11.7) \end{aligned}$ | 3.64 br.s ${ }^{\text {c }}$ | 4.14 s |
| arom | $\begin{aligned} & 6.63 \mathrm{~d}(8), 6.75- \\ & 7.15 \mathrm{~m}, 7.55 \mathrm{~d}(2) \\ & 7.73 \mathrm{dd}(8,2) \\ & 5.96,6.04 \mathrm{~d}(1.2) \end{aligned}$ | $7.38 \mathrm{~d}(2), 7.54 \mathrm{dd}(2,8)$ | $6.86-7.01 \mathrm{~m}$ | $6.83-7.00 \mathrm{~m}$ |
|  |  | $6.98 \mathrm{~d}(2), 6.86 \mathrm{dd}(2,8)$ | 7.62 d (2) | $7.60 \mathrm{~d}(2)$ |
|  |  | 6.77-6.81 | $7.47 \mathrm{dd}(8.5,2)$ | $7.73 \mathrm{dd}(2,8.5)$ |
| $\mathrm{OCH}_{2} \mathrm{O}$ or $\mathrm{OCH}_{3}$ |  | 5.96, 6.04d(1.2) | 3.81, 3.91, | 3.89,3.91, |
|  | $5.96,6.04 \mathrm{~d}(1.2)$ |  | $3.94,3.95 \mathrm{~s}$ | $3.94,3.95 \mathrm{~s}$ |
| OAc | - | $1.65,2.00 \mathrm{~s}$ | - | 1.70 s |
| OH | Obscured | - | 1.56, 1.96 | 2.2 |

${ }^{2}$ All the assignments are supported by appropriate spin decoupling experiments.
${ }^{\text {b }}$ Coupling constants in Hz .
${ }^{\text {'Sharpened after adding }} \mathrm{D}_{2} \mathrm{O}$.
The second new compound (2a), $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{8}, \mathrm{mp} 70^{\circ},[\alpha]^{28} \mathrm{D}+63.05\left(\mathrm{CHCl}_{3}\right)$, exhibited uv maxima at 231,276 , and 310 nm ( $\log \epsilon 4.06,3.80$, and 3.47). It formed a 2,4 -dinitrophenylhydrazone, mp $212-213^{\circ}$, but gave only a monoacetate (2b) on acetylation. However, the ir spectrum of $\mathbf{2 b}$ still contained a prominent OH peak at $3450 \mathrm{~cm}^{-1}$, indicating that $\mathbf{2 a}$ contains two hydroxyl groups only one of which can be readily acetylated. The ir spectra of both $\mathbf{2 a}$ and $\mathbf{2 b}$ also contained a prominent $\mathrm{C}=\mathrm{O}$ peak at $1660 \mathrm{~cm}^{-1}$, and the ${ }^{1} \mathrm{H}-\mathrm{nmr}$ and mass spectra gave clear evidence for the presence of two 3,4-dimethoxyphenyl groups. The presence of two low field aromatic protons at $\delta 7.74 \mathrm{dd}(8.5,2)$ and $\delta 7.62 \mathrm{~d}$ (2) suggested that one of the aryl groups was again present as an aroyl unit, and this was supported by the ms , which gave as the base peak an ion having $m / z 165\left(\mathrm{Ar}^{1} \mathrm{CO}^{+}\right)$. In contrast to the ${ }^{1} \mathrm{H}-\mathrm{nmr}$ spectrum of arborone (1a) and its acetate (1b), 7 -oxo-dihydrogmelinol (2a) and its acetate ( $\mathbf{2 b}$ ) did not display a high field signal corresponding to $\mathrm{H}-3$ but instead gave a one-proton triplet at $\delta$ 4.30 corresponding to $\mathrm{H}-4$. Both the multiplicity and the chemical shift of this signal place it unequivocally at $\mathrm{C}-4$ adjacent to the $\mathrm{C}=\mathrm{O}$ and $\mathrm{CH}_{2}$ groups. This observation together with the evidence for an isolated primary alcohol $\left(\mathrm{CH}_{2} \mathrm{OH}\right)$ group lead to the assignment of the structure $\mathbf{2 a}$, which is again supported by the mass spectrum.

## EXPERIMENTAL

General experimental procedures.-Ir spectra were recorded on a Shimadzu AR408 and uv spectra on a Beckman instrument. Mass spectra were recorded on an AEI MS9 spectrometer and 100 MHz
nmr spectra on an XL100 instrument. 360 MHz nmr spectra were provided by the Edinburgh University high field nerr service using a Bruker WH360 instrument.

Isolation of lignans.-The heartwood of G. arborea L. $(2,3)$ was obtained from Berhampur, Orissa State. The wood shavings ( 10 kg ) were powdered, and successively extracted with hexane and MeOH , and worked out following the procedure reported by Row et al. (2). The only operational modification was that no alkali was used at any stage.

The uncrystallizable gum ( 45 g ) obtained from the MeOH extract was absorbed on silica gel ( 100 g , $\mathrm{BDH}, 200$ mesh ) and chromatographed using a column of silica gel ( 800 g ) eluting successively with hexane ( 10 liters), hexane- $\mathrm{C}_{6} \mathrm{H}_{6}$ ( $9: 1,40$ liters; $1: 1,30$ liters), $\mathrm{C}_{6} \mathrm{H}_{6}$ ( 20 liters), and $\mathrm{C}_{6} \mathrm{H}_{6}-\mathrm{EtOAc}(9: 1,15$ liters). The elution was monitored by tlc and 1 -liter fractions collected. The compounds isolated are listed in Table 2. Methyl trans- $p$-methoxycinnamate, trans- $p$-hydroxycinnamic acid, paulownin acetate (8), and epieudesmin $(9,10)$ were isolated and identifed by mp, ir, and elemental analysis data.

Table 2. Data on Isolated Compounds

| Eluent | Fractions | Weight | Isolated Compounds |
| :---: | :---: | :---: | :---: |
| $\begin{gathered} \text { hexane } \\ \text { hexane- } \mathrm{C}_{6} \mathrm{H}_{6}(9: 1) \end{gathered}$ | 1-10 | 250 mg | oil |
|  | 11-13 | 30 mg | methyl $p$-methoxycinnamate |
|  | 14-30 | 150 mg |  |
|  | 31-40 | 150 mg | 2-O-ethylarboreol |
|  | 41-50 | 240 mg | 2-0-ethylarboreol |
| hexane- $\mathrm{C}_{6} \mathrm{H}_{6}(1: 1)$. | $51-60$ | 800 mg | 2-O-methylarboreol ( $350 \mathrm{mg} \mathrm{)}$ |
|  | 61-65 | 600 mg | paulownin acetate ( 700 mg ) $\beta$-sitosterol |
|  | 66-70 | 350 mg | paulownin |
|  | 71-72 | 400 mg | epieudesmin |
|  | 73-75 | 160 mg \} | epieudesmin ( 100 mg ) |
|  | 76-80 |  | arboreol ( 560 mg ) |
|  |  |  | isoarboreol ( 400 mg ) |
|  |  |  | arborone ( 25 mg ) |
| $\mathrm{C}_{6} \mathrm{H}_{6}$ | 81-90 | 7 g | arboreol, isoarboreol |
|  | 91-100 | 1.37 g | gmelinol, arboreol, isoarboreol |
| $\mathrm{C}_{6} \mathrm{H}_{6}-\mathrm{EtOAc}(9: 1)$ | 101-110 | 7.15 g | gmelinol ( 7 g ) |
|  | 111-115 | 560 mg | $p$-hydroxycinnamic acid ( 150 mg ) <br> gmelinol ( 530 mg ) <br> 7-oxo-dihydrogmelinol ( 20 mg ) |

Arborone (1a).-Crystallized from ErOAc as colorless needles mp $216-218^{\circ} ;[\alpha]^{28} \mathrm{D}+81.48(c$, $0.054, \mathrm{CHCl}_{3}$ ) and gave a positive Labat test for the $\mathrm{OCH}_{2} \mathrm{O}$ group; (Found: $\mathrm{C}, 62.10 ; \mathrm{H}, 4.75$. $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}_{8}$ requires $\mathrm{C}, 62.18 ; \mathrm{H}, 4.7 \%$ ); uv $\lambda \mathrm{MeOH} 206,234,285 \mathrm{~nm}(\log \in 4.44,4.17,4.09$, ; ir ( KBr ) $3515,3400,2900,1640(\mathrm{Ar}-\mathrm{C}=\mathrm{O}), 1600 \mathrm{~cm}^{-1}$; ms m/z $338\left(8 \%, \mathrm{M}_{-} \mathrm{H}_{2} \mathrm{O}-\mathrm{CH}_{2} \mathrm{O}\right), 192(2), 176$ (13), 164 (5), 161 (18), 150 (11), 149 (100), 135 (7), 131 (12), 103 (4), 64 (15), 63 (17).

Arborone diacetate (1b).-Arborone ( 7 mg ) was dissolved in 0.3 ml of dry pyridine, and 0.5 ml of dry $\mathrm{Ac}_{2} \mathrm{O}$ was added. The reaction mixture was kept at room temperature for 24 h . After the usual work up, the acetate was obtained as a gum that was purified on a small column of silica gel by eluting with $\mathrm{CHCl}_{3}$, when the acetate was obtained as a colorless oil ( 8 mg ); $\mathrm{Rf} 0.64\left(\mathrm{C}_{6} \mathrm{H}_{6}\right.$ - $\mathrm{EtOAc}, 9: 1$ ); Found: $\mathrm{C}, 61.2 ; \mathrm{H}$, 4.8. $\mathrm{C}_{24} \mathrm{H}_{22} \mathrm{O}_{10}$ requires $\mathrm{C}, 61.27 ; \mathrm{H}, 4.72 \%$ ); uv $\lambda(\mathrm{MeOH}) 234,284,313 \mathrm{~nm}(\log \in 4.44,4.17,4.11)$; ir $\left(\mathrm{CHCl}_{3}\right) 1740$ (acetate $\left.\mathrm{C}=\mathrm{O}\right), 1675,1610,1510 \mathrm{~cm}^{+1}$; ms $m / z 470\left(1 \%, \mathrm{M}^{+}\right), 351(7), 350(18), 338$ (3), 337 (15), 261 (3), 228 (3), 202 (3), 201 (3), 194 (3), 177 (4), 176 (6), 171 (2), 161 (2), 150 (11), 149 (100), 135 (6), 121 (13), 65 (14).

Conversion of Arborone to Arboreol (3). -Arborone ( 5 mg ) was dissolved in EtOAc ( 1 ml ), and to it HCl $(0.5 \mathrm{ml})$ was added. The solution was kept at room temperature for 24 h and worked up in the usual manner. The light yellow residue on crystallization from $\mathrm{C}_{6} \mathrm{H}_{6}$ furnished colorless needles ( $80 \%$ ), mp $162^{\circ}$, $[\alpha]^{28} \mathrm{D}+75^{\circ}$, Rf $0.25\left(\mathrm{C}_{6} \mathrm{H}_{6}\right.$ - $\left.\mathrm{EtOAc}, 9: 1\right)$. The mmp was unchanged with an authentic sample of arboreol.

7-Oxo-dihydrogmelinol (2a).-Crystallized from $\mathrm{C}_{6} \mathrm{H}_{6}$ /petroleum ether ( $40-60^{\circ}$ ) as colorless needles, $\mathrm{mp} 70^{\circ} ;[\alpha]^{28} \mathrm{D}+63.05 i,\left(1.134, \mathrm{CHCl}_{3}\right)$; (Found $\mathrm{C}, 63.10 ; \mathrm{H}, 6.36 . \mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{8}$ requires $\mathrm{C}, 63.15 ; \mathrm{H}$, $6.27 \%)$; uv $\lambda(\mathrm{MeOH}) 231,276,310 \mathrm{~nm}(\log \in 4.06,3.8,3.47)$; ir $(\mathrm{KBr}) 3500,2950,12660(\mathrm{Ar}-\mathrm{C}=\mathrm{O})$,

1595, $1518 \mathrm{~cm}^{-1}$; ms $m / z 382\left(10 \%, \mathrm{M}_{\left.-2 \mathrm{H}_{2} \mathrm{O}\right), 192(30), 167(35), 166(14), 165(100), 151(7), 139}\right.$ (29), 124 (11), 122 (11), 108 (10), 107 (10), 95 (10), 94 (10), 92 (12), 79 (38), 77 (32).

7-Oxo-dibydrogmelinol acetate (2b).-Treatment of 2a with $\mathrm{Ac}_{2} \mathrm{O} /$ pyridine gave a colorless oil (7 mg); Rf $0.37\left(\mathrm{C}_{6} \mathrm{H}_{6}\right.$-EtOAc, 3:2); (Found: $\mathrm{C}, 62.54 ; \mathrm{H}, 6.2$. $\mathrm{C}_{24} \mathrm{H}_{28} \mathrm{O}_{9}$ requires $\mathrm{C}, 62.6 ; \mathrm{H}, 6.13 \%$ ); uv $\lambda$ $(\mathrm{MeOH}) 231,281,308 \mathrm{~nm}(\log \in 4.47,4.28,4.11)$; ir $\left(\mathrm{CHCl}_{3} 3450(\mathrm{OH}), 1735\right.$ (acetate $\mathrm{C}=\mathrm{O}$ ), 1655, $1590,1510 \mathrm{~cm}^{-1} ; \mathrm{ms} \mathrm{m} / \mathrm{z} 443$ ( $3 \%, \mathrm{M}-\mathrm{OH}$ ), $384(8), 383$ (13), 382 (46), 381 (9), 370 (6), 368 (9), 360 (7), 352 (7), 351 (28), 339 (6), 295 (6), 278 (24), 277 (84), 276 (7), 269 (8), 268 (36), 250 (9), 240 (28), 235 (10), 234 (13), 233 (13), 232 (17), 217 (18), 209 (22), 208 ( 67 ), 192 (35), 167 (60), 166 (18), 139 (44), 137 (13), 124 (11), 122 (14), 107 (12).

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